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CHAPTER 13

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and Colligative

Properties SECTION

1 SHORT ANSWER

Answer the

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Chemistry

following questions
in the space
provided. 1. Use
the guidelines in
Table 1 on page
437 of the text to
predict the
solubility of the
following
compounds in
water: soluble a.
magnesium nitrate
insoluble b. barium
sulfate insoluble c.

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calcium carbonate

Section 13 1

13 Ions in Aqueous
Solutions and

Colligative

Properties

Chapter 13 Review

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Teacher Notes and

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12 SECTION 1

SHORT ANSWER 1.

c 2. a 3. b 2. a.

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alcohol b. water c.
the gels 3. The
mixture is a colloid.
The properties are
consistent with
those reported in
Table 3 on page
404 of the text.
The particle size is
small, but not too
small, and the
mixture

CHAPTER 12

Page 17/32

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SECTION 13-2

SHORT ANSWER

Answer the following questions in the space provided. 1.

Following are statements about the dissolving process. Explain each one at the

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molecular level. a.
Increasing the
pressure of a
solute gas above a
liquid solution
increases the
solubility of the gas
in the liquid. b.

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5 Test Worksheet 2

Name _____ Per _____

Find the solutions

to the system of

equations. Write

the answers in

coordinate form. 1.

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2.3. Compare each graph to the parent function $y = x^2$.

Determine the transformations for each of the functions.

Review for Section

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Review for ...

Also, adding 1.0 L of water to the solid

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does not produce
1.0 L of solution. He
did not make a
0.600 M solution. b.
Mary needs a 0.02
M solution of
 NaNO_3 □ She
calculates that she
needs 2.00 g of
 NaNO_3 for 0.02
mol. She puts this
solid into a 1.00 L
volumetric flask
and fills the flask to

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the 1.00 L mark.

Section 13.1

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problem 1 from AY

2019-2020 HW 7&8

before solving this

problem. Write

code in a

programming

language of your

choice to generate

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Work all of Chapter
13 Review

Questions (whether
they are assigned

or not). Work

through all of the
problems before

looking at the

answers, and then

correct each of the

problems. The

entire solution is

shown in the

answer section at

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the back of the
text.

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