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The midterm exam
covers chapters 1 - 4 & 9 -

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11. You should read through each chapter, look over old tests you still have. answer the following questions and do the calculations in order prepare yourself for the mid-term. 1. Define the following terms and describe where each is located.

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Chemistry - Mid Term
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The midterm exam covers chapters 1 - 4 & 9 - 11. You should read through each chapter, look over old tests you still have. answer the following questions and do the calculations in order prepare yourself for the mid-term. 1.

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terms
Answers

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Spell. Test. PLAY.
Match. Gravity. Created
by. nataliedrumm. Chem
I vocabulary. Terms in
this set (91) chemical
bond. a mutual electrical

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attraction between the nuclei and valence electrons of different atoms that binds the atoms together.

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answers. Draw an
electron dot structure for

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the following then predict the oxidation number or charge: a. Nitrogen b. argon c. magnesium d. sodium e. aluminum f. sulfur g. silicon h. fluorine 66. Use the electronegativity to determine what type of bond will form for the following atoms.

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Chapters 2-11 (excluding
heat and calorimetry in
chapter 3). 50 Questions
/ 50 points total Each
question has only 1
correct answer Honors
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ALL ANSWERS ON A
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Sheet #1 The midterm
exam covers chapters I -
4 & 9 - I I You should

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read through each chapter, look over old tests you still have answer the following questions and do the calculations in order prepare yourself for the mid-term | Define the following terms and describe where each is located proton - neutron -

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Midterm: Review Packet

1. Write in the number of significant figures for each of the following measurements: _____ 3.00

mL _____ 0.00290 g _____

50.00 m _____ 0.070 kg

_____ 400 L 2. Using the

following masses: 0.0562

kg, 124.213 g and 1635

mg. The answer should

be reported as: _____ g

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Honors Chemistry

Midterm: Review Packet

Chemistry Midterm

Review ANSWERS 1

Problem Solving 1. A

piece of ice has a mass of
9.2 g and a piece volume
of 10.0 cm³. volume a.

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Chemistry Midterm

Exam Review Sheet

Spring 2012. 1. Know

your safety rules. 2.

shopping mall wanted to

determine whether the

more expensive “ Tough

Stuff ” floor wax was

better A. than the

cheaper “ Steel Seal ”

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Answers

floor wax at protecting its floor tiles against scratches. One liter of each brand of floor wax was applied to each of 5 test sections of the main hall of the mall.

Chemistry Midterm
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Exam Review Sheet #1
The midterm exam
covers chapters 1 - 4 & 9 -

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11. You should read through each chapter, look over old tests you still have. answer the following questions and do the calculations in order prepare yourself for the mid-term. 1. Define the following terms and describe where each is located. proton - neutron -

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Guadagna
Chemistry - Mid Term
Exam Review Sheet #1

The midterm exam covers chapters 1 - 4 & 9 - 11. You should read through each chapter, look over old tests you still have. answer the following questions and do the calculations in order prepare yourself for the mid-term. I.

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Define the following terms and describe where each is located. proton - neutron -

Chemistry - Mid Term Exam Review Sheet #1
answer choices The distance between the gas particles is much greater than the size of the particles As the gas particles collide, the total energy of the system

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Answers

increases The gas particles have strong intermolecular forces

Regents Chemistry

Midterm Review Quiz - Quizizz

In chemistry classes, we get to discuss a lot when it comes to liquid and how to form a solution. With the midterms just around the corner, it is becoming more critical

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to ensure that you are up to date with all that has been covered in class so far. Below is a chemistry mid-term exam practice test designed to refresh your memory. Give it a shot!

Chemistry Exam Practice
Test! Trivia Quiz
Questions ...
Chemistry 2 Midterm
Review Answers

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CHAPTER 1 1. a.

mixture, homogeneous

b. mixture,

heterogeneous (if
particles in liquid) c. pure
substance, element d.

mixture, probably

heterogeneous but would
not be determined by
visual inspection alone e.

. pure substance, element

f. pure substance,

compound g. mixture,

heterogeneous

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