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~~Periodic~~

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~~explanation, QA~~

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Electrons in

Atoms Although

the speed of all

electromagnetic

waves in a

vacuum is the

same, waves can

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have different wavelengths and frequencies. As you can see from the equation on the previous page, wavelength and frequency are inversely related; in other words, as one quantity increases, the other decreases.

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each electron
occupies the
lowest energy

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available, it is fundamentally impossible to know precisely both the velocity and position of a particle each electron occupies the lowest energy orbital available.

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Q. The Quantum Mechanical Model of the Atom describes the electron's probable location around the nucleus in a

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3-D cloud called
a(n) _____.

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Electrons are found in certain orbits located around the nucleus. Every electron has a fixed energy in certain energy levels: they are said to be quantized. Electrons

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farther away
from the nucleus
have higher
energy than
electrons closer
to the nucleus.

Energy levels

are closer
together the

farther away

from the

nucleus.

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Terms in this
set (41)

periodic table.
an arrangement
of the elements
in order of
their atomic
numbers so that
elements with
similar
properties fall

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in the same
column, or
group. ...

Electrons are
added as well,
but ...

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The orbital
diagram for a

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ground-state

nitrogen atom is
 $1s^2 2s^2 2p^3$ A: 5.

The number of
orbitals in a d
subshell is A: 5
6. The maximum
number of
electrons that
can occupy an
energy level
described by the
principal
quantum number,

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Chapter 5: $2n^2$ 7.

A ground-state
atom of

manganese has

___ unpaired

electrons and is

_____. A: 5,

paramagnetic 8.

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~~Electrons in an~~

~~orbital with ...~~

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Answer. The

atomic mass of

chlorine is very

far from a whole

because a

weighted average

of atomic masses

of all of its

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isotopes is

computed in
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determining its
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atomic mass.

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Step.

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two electrons
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electrons

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and waves. d.

the chemical

properties of

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Elements can be
grouped
according to
periodicity but
physical
properties

cannot. _____ 28.

Elements in a
group or column
in the periodic
table can be
expected to have
similar. ...

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resonance is a
way of
describing
delocalized
electrons in an

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atom or molecule

that cannot be
represented with
a single Lewis

dot structure.

... · Determine
the total # of
valence

electrons in a
molecule: N 5.

30 18. Ve ...

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