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Homogeneous Catalysis. As the name implies, homogeneous catalysts are present in the same phase (gas or liquid solution) as the reactants. Homogeneous catalysts generally enter directly into the chemical reaction (by forming a new compound or complex with a reactant), but are released in their initial form after the reaction is complete, so that they do not appear in the net reaction equation.

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A new highly active and selective catalyst for the synthesis of  $\beta$ -lactones from CO and epoxides is reported. The catalyst, [(N,N)-bis(3,5-di-tert-butylsilylylidene) phenylene]diamino)Al(THF)<sub>2</sub>][Co(CO)<sub>4</sub>] ((salph)Al(THF)<sub>2</sub>][Co(CO)<sub>4</sub>] is easily prepared from the corresponding (salph)AlCl and NaCo(CO)<sub>4</sub>. At 50 °C and 880 psi of CO, the catalyst (1 mol %) carbonylates epoxides such as propylene ...

**Synthesis of  $\beta$ -Lactones: A Highly Active and Selective**

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For this experiment, TLC is used to monitor the reaction of the reacti on solution, containing 2-naphthol, benzyl-tri- n -butylammonium chloride (the catalyst), and allyl bromide, in dichloromethane. At 0 minutes, 10 minutes, 30 minutes, and 60 minutes, samples from the bottom organic phase of the mixture are taken. The R.

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